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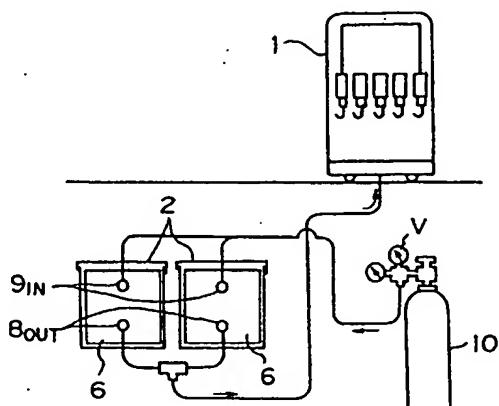
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(54) Bag for bag-in-box.

(57) A bag accommodated in an outer box of a  
bag-in-box is made of film-like material to be  
collapsible to push out the contents such as  
beverage syrups and liquid foods and com-  
prises a bag body and a mouthpiece provided  
on the bag body for passing the contents therethrough. The bag body is made of polyester resin  
at a thickness of 20 - 150 µm.

F I G. 1



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BACKGROUND OF THE INVENTION

This invention relates to an inner bag accommodated in an outer box of a bag-in-box.

5 A bag-in-box is a type of package in which a collapsible bag is accommodated in an outer box for protection in order to transport and store liquid. The bag-in-box is often used because it is light in weight, convenient for transportation and relatively inexpensive in comparison with a conventional bottle, can or tank.

10 However, if a conventional plastic bag, especially polyethylene bag is formed thinner, gas permeability is getting relatively higher (gas barrier property is getting lower) and it causes to deteriorate quality of contents such as liquid due to oxygen. In order to prevent the deterioration, a plastic bag laminated with other films is generally used. Further, polyethylene has a property of absorbing smell components in environments, so it is known that smell or aroma sticks to contents in the bag. The conventional bag is not strong enough to accommodate liquid including carbon dioxide.

SUMMARY OF INVENTION

15 It is an object to provide a bag for a bag-in-box, which maintains gas barrier property and pressure resistance and can prevent absorption and elution of smell or aroma component, and to enable the bag to be used for accommodating beverage syrup food which a conventional bag cannot accommodate.

20 According to the invention, there is provided a bag accommodated in an outer box of a bag-in-box, which is made of film-like material so as to be collapsible to push out contents therein such as beverage syrups and liquid foods, comprising a collapsible bag body made of polyester resin at a thickness of 20 - 150 µm, preferably 50 - 125 µm, and a mouthpiece provided on the bag body for passing the contents therethrough.

25 Other objects and advantageous effects of this invention will be explained in detail with respect to the accompanying drawings mentioned below.

BRIEF DESCRIPTION OF THE DRAWINGS

In the accompanying drawings:

30 Figure 1 is a structural view of a dispenser system;  
 Figure 2 is a vertically sectional view of the bag-in-box;  
 Figure 3 is a perspective view of an inner bag for a bag-in-box according to this invention; and  
 Figure 4 is an inner structural view of a temperature controlled box for preservation.

DETAILED DESCRIPTION OF THE INVENTION

35 First, a dispensing system for dispensing cooling beverage such as juice, cola, etc. and a drink such as coffee and tea by using a bag-in-box with a dispenser will now be explained.

40 In Figures 1 and 2, a dispenser 1 is connected to two bag-in-boxes 6 and 6 accommodated in two pressure resistance boxes 2 and 2 and a carbon dioxide tank or cylinder 10 is also connected to the bag-in-boxes 6 and 6 in the pressure resistance box 2 through a pressure-reducing valve V.

45 Each bag-in-box 6 comprises, as shown in Figure 2, an outer box 7 made of corrugated cardboard, and two upper and lower inner bags 8 and 9 accommodated vertically adjacent to each other in the outer box 7. The lower inner bag 8 is for accommodating syrup for drinking and the outlet 8 out is connected to the dispenser 1. The inlet 9 IN of the upper inner bag 9 is connected to the carbon dioxide tank 10 as a pressure source. The lower inner bag 8 has a bag body 17 for containing contents such as cooling beverages and a mouthpiece 12 for pouring out the contents as shown in Figure 3.

The bag body 17 is made of polyester resin, that is, polyethylene terephthalate resin, polyethylene naphthalate resin and copolymerized polyester resins (1) - (7) mentioned below in detail.

Each resin will now be explained in more detail.

Polyethylene terephthalate resin

55 Polyethylene terephthalate resin (PET) for the bag body 17 according to this invention is made of terephthalic acid or its ester forming derivative and ethylene glycol or its ester forming derivative. The polyethylene terephthalate resin may be copolymerized with other dicarboxylic acid of 20 mol% or less and/or dihydroxy compound less of 20 mol% or less. As dicarboxylic acid used for copolymerization in addition to terephthalic acid, aromatic dicarboxylic acid such as phthalic acid, isophthalic acid, naphthalene dicarboxylic acid, diphenyl dicarboxylic acid and diphenoxethane dicarboxylic acid; or aliphatic dicarboxylic acid such as adipic acid, se-

basic acid, azelaic acid and decanedioic acid; or alicyclic dicarboxylic acid such as cyclohexane dicarboxylic acid may be used.

Further, as dihydroxy compound used for copolymerization in addition to ethylene glycol, aliphatic glycol such as trimethylene glycol, propylene glycol, tetramethylene glycol, neopentyl glycol, hexamethylene glycol and dodecamethylene glycol; or alicyclic glycol such as cyclohexane dimethanol; or bisphenol group; or aromatic diol group such as hydroquinone and 2,3-bis (4- $\beta$  hydroxyethoxyphenyl) propane.

Such polyethylene terephthalate resin forms substantially linear polyester in such a manner that only ethylene terephthalate component unit itself is ester-bonded or ethylene terephthalate component unit and dioxyethylene terephthalate component unit are disposed at random to be ester-bonded. It is confirmed due to the fact that the polyethylene terephthalate resin dissolves in o-chlorophenol that the polyethylene terephthalate resin is substantially linear.

The limiting (intrinsic) viscosity [ $\eta$ ] of polyethylene terephthalate resin (value obtained at 25°C in o-chlorophenol) is normally 0.6 - 1.5 dl/g, preferably 0.7 - 1.2 dl/g. The melting point thereof is normally 210°-265°C, preferably 220 - 260°C and the glass transition point thereof is normally 50 - 120°C, preferably 60 - 100°C.

#### Polyethylene naphthalate resin

Polyethylene naphthalate resin used for the bag body 17 according to this invention includes ethylene-2,6-naphthalate structural unit, derived from 2,6-naphthalate dicarboxylic acid and ethylene glycol, of 60 mol% or more, preferably 80 mol% or more, more preferably 90 mol% or more. However, the polyethylene naphthalate resin may include a structural unit less than 40 mol% other than ethylene-2,6-naphthalate.

As structural unit other than ethylene-2,6-naphthalate, structural units derived from the following components (A group + B group may be selected.

A group: aromatic dicarboxylic acid such as terephthalic acid, isophthalic acid, 2,7-naphthalene dicarboxylic acid, 1,7-naphthalene dicarboxylic acid, diphenyl-4,4'-dicarboxylic acid, 4,4'-diphenylether dicarboxylic acid, 4,4'-diphenylsulfone dicarboxylic acid, 4,4'-diphenoxylethane dicarboxylic acid and dibromoterephthalic acid; aliphatic dicarboxylic acid such as adipic acid, azelaic acid, sebacic acid and decanedioic acid, alicyclic dicarboxylic acid such as 1,4-cyclohexand dicarboxylic acid, cyclopropane dicarboxylic acid, hexahydro terephthalic acid, and hydroxy carboxylic acid such as glycolic acid, p-hydroxy benzoic acid, p-hydroxy ethoxy benzoic acid.

B group: propylene glycol, trimethylene glycol (1,3-propanediol), diethylene glycol, tetramethylene glycol, pentamethylene glycol, hexamethylene glycol, decamethylene glycol, neopentylglycol, p-xylene glycol, 1,4-cyclohexane dimethanol, bisphenol A, p,p-diphenoxysulfone, 1,4-bis ( $\beta$ -hydroxyethoxy) benzene, 2,2-bis (p- $\beta$ -hydroxy ethoxyphenyl) propane, polyalkylene glycol, p-phenylene bisdimethyl siloxane and glycerol.

Further, the polyethylene naphthalate resin used in this invention may include a little structural unit of, e.g. 2 mol% or less, derived from polyfunctional compound such as trimethylol ethane, trimethylol propane and trimethylol methane. The polyethylene naphthalate resin used in this invention may include a little structural unit of, e.g. 2 mol% or less, derived from single functionality compound such as benzoylbenzoic acid, diphenylsulfone-monocarboxylic acid, stearic acid, methoxypolyethylene glycol and phenoxy polyethylene glycol. Such polyethylene naphthalate resin is substantially linear. This is confirmed due to the fact that the polyethylene naphthalate resin dissolves in o-chlorophenol. The limiting viscosity [ $\eta$ ] of polyethylene naphthalate resin at a temperature of 25°C in o-chlorophenol is 0.2 - 1.1 dl/g, preferably 0.3 - 0.9 dl/g, more preferably 0.4 - 0.8 dl/g.

The limiting viscosity [ $\eta$ ] of the polyethylene naphthalate resin is measured by the following method. That is, the polyethylene naphthalate resin is dissolved at a density of 1g/100 ml, and viscosity of the solution is then measured by Ubbelohde type capillary viscometer. Thereafter, o-chlorophenol is gradually added to the solution to measure the viscosity of solution on the side of a low density. Finally, the limiting viscosity [ $\eta$ ] is obtained through extrapolation.

The temperature (Tc) of crystallization of polyethylene naphthalate resin at the time of raising the temperature thereof at a rate of 10°C/minute by a differential scanning calorimeter (DSC) is normally 150°C or more, preferably 160 - 230°C, more preferable 170 - 220°C. The temperature (Tc) thereof is measured by the following method. That is, the polyethylene naphthalate resin is dried for approximately more than 5 hours at a temperature of approximately 140°C under a pressure of approximately 5 mmHg to prepare a thin piece of 10 mmg as a sample by cutting the dried resin. The sample is put into an aluminium pan for liquid in an atmosphere of nitrogen to measure the temperature of crystallization thereof by means of a DSC-2 type differential scanning calorimeter manufactured by PERKIN ELMER CO.. The measuring conditions are as follows. First, the tem-

perature of the sample is raised rapidly to 290°C from a room temperature and is maintained at a temperature of 290°C for 10 minutes to melt the sample. Thereafter, the sample is rapidly cooled down to a room temperature. Then, the temperature of the sample is increased at a rate of 10°C/minute. Under this condition, a temperature at the time when heat generation becomes the maximum value is measured. The temperature at that time is the temperature of crystallization at time of raising temperature.

#### Copolymerized polyester resin (1)

The copolymerized polyester resin (1) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit and isophthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit.

The terephthalic acid component unit is included therein by 85 - 99.5 mol%, preferably 90 - 99.5 mol% and the isophthalic acid component unit is included therein by 0.5 - 15 mol%, preferably 0.5 - 10 mol%. The copolymerized polyester resin may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the terephthalic acid and the isophthalic acid mentioned above as a component of the dicarboxylic acid. Such dicarboxylic acid is, for example, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (1) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the ethylene glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol, 1,4-butanediol, neopentyl glycol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyethoxyphenyl) propane or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

#### Copolymerized polyester resin (2)

The copolymerized polyester resin (2) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit and 2,6-naphthalene dicarboxylic acid component unit; and dihydroxy compound structural unit including ethylene glycol component.

The terephthalic acid component unit is included therein by 80 - 99.5 mol%, preferably 90-99.5 mol% and 2,6-naphthalene dicarboxylic acid component unit is included therein by 0.5 - 20 mol%, preferably 0.5 - 10 mol%. The copolymerized polyester resin (2) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the terephthalic acid and the 2,6-naphthalene dicarboxylic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid or 2-methylterephthalic acid. Further, the copolymerized polyester resin (2) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the ethylene glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propane diol, 1,4-butanediol, neopentyl glycol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxy) benzene, 1,4-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyphenyl) propane or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

#### Copolymerized polyester resin (3)

The copolymerized polyester resin (3) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit and adipic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit.

The terephthalic acid component unit is included therein by 85 - 99.5 mol%, preferably 90 - 99.5 mol% and the adipic acid component unit is included therein by 0.5 - 15 mol%, preferably 0.5 - 10 mol%. The copolymerized polyester resin (3) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (3) does not lose its property in addition to the terephthalic acid and the adipic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (3) according to this invention may include other dihydroxy compound of 1 mol% or less with a range of quantity at which the copolymerized polyester resin (3) does not lose its property in addition to the ethylene glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol, 1,4-butanediol, neopentyl glycol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxy) benzene, 1,4-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyethoxyphenyl) propane or bis (4- $\beta$ -hydroxy-

thoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

#### Copolymerized polyester resin (4)

5 The copolymerized polyester resin (4) for the main body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and diethylene glycol component unit.

10 The ethylene glycol component unit is included therein by 93-98 mol%, preferably 95-98 mol% and the diethylene glycol component unit is included therein by 2-7 mol%, preferably 2-5 mol%. The copolymerized polyester resin (4) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (4) does not lose its property in addition to the terephthalic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (4) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of 15 quantity at which the copolymerized polyester resin (4) does not lose its property in addition to the diethylene glycol and the diethylene glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol; 1,4-butanediol, neopentylglycol, cyclohexanediol, cyclohexanedimethanol, 1,3 bis (2-hydroxyethoxy) benzene, 1,4 bis (2-hydroxyethoxy) benzene, or bis (4- $\beta$ -hydroxyethoxyphenyl) propane, or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

#### Copolymerized polyester resin (5)

20 The polymerized polyester resin (5) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit and isophthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and neopentyl glycol component unit.

25 The ethylene glycol component unit is included 85-99.5 mol% preferably 90-99.5 mol% and the neopentyl glycol of 0.5-15 mol%, preferably 0.5-10 mol%. The copolymerized polyester resin (5) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the terephthalic acid and the isophthalic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (5) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (5) does not lose its property in addition to the ethylene glycol and neopentyl glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol, 1,4-butanediol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxyphenyl) benzene, 1,4-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyphenyl) propane or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

#### Copolymerized polyester resin (6)

40 The copolymerized polyester resin (6) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit including terephthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and cyclohexane dimethanol component unit.

45 The ethylene glycol component unit is included therein by 85-99.5 mol%, preferably 90-99.5 mol% and the cyclohexane dimethanol component is included therein by 0.5-15 mol%, preferably 0.5-10 mol%. The copolymerized polyester resin (6) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (6) does not lose its property in addition to the terephthalic acid and the isophthalic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (6) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (6) does not lose its property in addition to the ethylene glycol and the cyclohexane dimethanol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol, 1,4-butanediol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxy) benzene, 1,4-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyethoxyphenyl) propane or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

Copolymerized polyester resin (7)

The copolymerized polyester resin (7) for the bag body 17 according to this invention comprises dicarboxylic acid structural unit dihydroxy compound structural unit and polyhydroxy compound structural unit including at least three hydroxy groups.

The copolymerized polyester resin (7) comprises dicarboxylic acid structural unit includes isophthalic acid component unit of 20-100 mol%, preferably 50-98 mol% and terephthalic acid component acid of 0-80 mol%, preferably 0.5-50 mol%. The dihydroxy compound structural unit may include dihydroxyethoxy resole component unit of 5-90%, preferably 10-85 mol%, or ethylene glycol component of 10-95 mol%, preferably 15-90 mol%.

The copolymerized polyester resin (7) may include polyhydric compound structural unit of 0.05-1.0 parts by mol, preferably 0.1-0.5 parts by mol with respect to dicarboxylic acid component unit of 100 parts by mol. Such polyhydric compound structural unit is derived from the compound of trimethylol methane, trimethylol ethane, and trimethylol propane. Among these components, trimethylol propane is preferable. The copolymerized polyester resin (7) may include other dicarboxylic acid of 1 mol% or less within a range of quantity at which the copolymerized polyester resin does not lose its property in addition to the terephthalic acid and the isophthalic acid mentioned above as a component of dicarboxylic acid. Such dicarboxylic acid is, for example, isophthalic acid, phthalic acid, 2-methylterephthalic acid or 2,6-naphthalene dicarboxylic acid. Further, the copolymerized polyester resin (7) according to this invention may include other dihydroxy compound of 1 mol% or less within a range of quantity at which the copolymerized polyester resin (7) does not lose its property in addition to the dihydroxyethoxy resole and the ethylene glycol mentioned above. Such dihydroxy compound is, for example, 1,3-propanediol, 1,4-butanediol, neopentyl glycol, cyclohexanediol, cyclohexanedimethanol, 1,3-bis (2-hydroxyethoxy) benzene, 1,4-bis (2-hydroxyethoxy) benzene, 2,2-bis (4- $\beta$ -hydroxyethoxyphenyl) propane or bis (4- $\beta$ -hydroxyethoxyphenyl) sulfone. Each compound includes carbon atoms of 3 to 15.

The molecular weight of the above copolymerized polyester resins (1) - (7) is not specially restricted if it is within a range in which various articles, that is, bags can be made of polyester resin composite, and, however, it is preferably that the limiting viscosity [ $\eta$ ] of copolymerized polyester resin in o-chlorophenol solvent is within 0.5 dl/g - 1.5 dl/g, preferably 0.6 dl/g - 1.2 dl/g.

The bag body 17 according to this invention is made of one kind of resin selected from the above various polyester resin. However, the bag body 17 may be made of laminated films of more than two kinds of polyester resin. In this case, polyamide resin may be used as an intermediate layer disposed between laminated layers. The above resins may include various additives such as thermal stabilizer, weather resistant stabilizer, antistatic agent, lubricant, inorganic filler, mould releasing agent, pigment dispersing agent, pigment of dye.

The thickness of the bag body 17 made of the above materials is within 20-150  $\mu\text{m}$ , preferably 50 - 126  $\mu\text{m}$ . If the thickness is above 150  $\mu\text{m}$ , the bag cannot be smoothly folded. If the thickness is under 20  $\mu\text{m}$ , the strength is not enough. These bags are manufactured by various resin forming methods, especially a blow moulding method in which a preform is heated to be softened in a heated mould, and compressed air is then supplied into the preform to expand it, so that the expanded body contacts the inner wall of the cavity of the mould. The expanded hollow body as a bag body is separated from the mould. A preform is usually manufactured by an injection moulding. However, the preform may be manufactured by extrusion moulding in which heated and melted plastic material is extruded out. A simple frame for adjusting the shape of the bag may be used without using a conventional mould. Moreover, a drawing blow moulding without the mold and the frame can be adopted.

In Figure 3, the mouthpiece 12 connected to the bag body 17 of the bag 8 may be made of the same material as that of the bag body 17, and may be formed as an assembled body in which metal is combined with the resin.

In case that the bag body 17 is manufactured through the blow moulding of the preform which is formed by an injection moulding, a volume expansion rate of the bag body 17 with respect to the volume of the preform is within 100 - 240, preferably 120 - 200. The volume expansion rate is defined as follows.

$$\text{Volume expansion rate} = \frac{\text{Inner volume of bag body}}{\text{Inner volume of preformed body}}$$

The bag body for a bag-in-box formed in this manner gradually shrinks as the contents filled in the bag are gradually taken out, and the shrinking direction of the bag is not specially restricted.

This invention will now be explained in more detail with reference to examples.

Example 1

## 1. Manufacture of sample 1 according to this invention.

5       A preform was produced in such a manner that polyethylene terephthalate resin [J125 manufactured by  
 MITSUI PET RESIN CO. (limited viscosity [ $\eta$ ] is 0.71 dL/g) was heated to a temperature of 270 - 290°C to be  
 injected. The preform was heated by an infra-red heater to 100°C, and was then extended or expanded freely  
 in a simple mould so that the thickness of the main body of a bag was 100 µm and the volume expansion rate  
 was 159. The bag (sample 1 according to this invention) was thus manufactured.  
 10

## 2. Preparation of compared samples 1 to 3.

A laminated bag mainly made of polyethylene available at present was prepared. The concrete laminated  
 layer structures were as follows.

15     Compared sample 1: PE (40 µm)/EVAL (10 µm) PE (40 µm) + PE (40 µm)/EVAL (10 µm) PE (40 µm)  
 Compared sample 2: PE (50 µm)/ONY (20 µm)/PE (50 µm) + ONY (20 µm)/PE (50 µm)  
 Compared sample 3: EVA (80 µm) + PE (30 µm)/KONY (20 µm)/PE (50 µm)  
 wherein PET is polyethylene terephthalate, PE is polyethylene, EVAL is ethylene-vinylalcohol copolymer, ONY  
 20     is oriented nylon, KONY is oriented nylon coated with vinylidene chloride, EVA is ethylene-vinylacetate copo-  
 lymer and OPP is oriented polypropylene. The following preservation test for mineral water was performed with  
 reference to the sample 1 of this invention and the three compared samples 1 to 3. First, mineral water was  
 filled into each bag while a heat-sterilization is performed. These bags with mineral water therein were put  
 into a temperature controlled box 20 of 30°C to be preserved therein at a relative humidity of 75% for eight  
 25     weeks as shown in Figure 4. The temperature controlled box 20 has therein a plurality of shelves 30, 30 --- 30  
 on which there are some compared samples 31, 31 --- 31, some invented samples 32, 32 --- 32 and some  
 smell generators 26 and 26 to stick the smell on each bag. At the bottom of the box 20 is provided an electric  
 heater 25 which is controlled by a temperature sensor 23 and a temperature control unit 22. An air circulating  
 fan 21 is provided at an upper portion of the box 20 to circulate the air in an accommodated space of the box  
 30     20, and the air circulating fan 21, the electric heater 21, the temperature control unit 22 and the temperature  
 sensor 23 are connected to a plug 24. The smell generator 26 included smell components such as i) citrus  
 series d-limonene as base, lemon terpene, and orange C<sub>6</sub> aldehyde, ii) coffee series flavone, and iii) lactic ser-  
 35     ies C<sub>10</sub> lactone, and diacetyl. Each smell component was diluted by ethanol to be mixed with each other at a  
 mixing rate of i):ii):iii)=5:2:1. Then, ten persons tried to smell each sample in bag. Each person judged the order  
 (ranking) of smell on each sample in bag. The result is shown in Table 1 in which each position means the level  
 of smell, and, for example, the first position means that the level of smell is lowest in comparison with other  
 bags and the fourth position means that the level of smell is highest in comparison with other bags.

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Table 1

	Heat filling immediately after	One week after	Two weeks after	Four weeks after	Eight weeks after
Invented sample 1	First position ** (10/10)	First position ** (10/10)	First position ** (10/10)	First position ** (10/10)	First position ** (10/10)
Compared sample 1	Second position * (8/10)	Second position (5/10)	Second position (6/10)	Second position (5/10)	Second position (6/10)
Compared sample 2	Third position (5/10)	Third position (4/10)	Third position (3/10)	Third position (5/10)	Third position (4/10)
Compared sample 3	Fourth position (6/10)	Fourth position (5/10)	Fourth position (8/10) *	Fourth position (5/10)	Fourth position (6/10)

\*; The level of significance is 5%

\*\*; The level of significance is 10%

The number in the parentheses means the number of the panel who ranked each position/total panel number.

Judging from Table 1, the invented sample 1 made of PET according to this invention keeps the first position during eight weeks; that is, the sample 1 could effectively prevent a smell from passing through the bag, so that the smell was not stuck to the contents of the bag.

Preservation test was performed with reference to the invented sample 1 in Example 1 and the compared sample 1 in Example 1. Each sample was heat-filled with citrus series syrup. The compared sample 1 was selected because it was the best one among conventional bags as a result of the above preservation test in Example 1. The preservation condition was the same as in Example 1. The triangle test was conducted by ten persons to distinguish invented sample 1 and compared sample 1. The result is shown in Table 2.

Table 2

	One week after	Two weeks after	Eight weeks after
Distinction between invented sample 1 and compared sample 1	5 persons were correct	8 persons * were correct	10 persons ** were correct
Person who felt invented sample 1 was better than compared sample 1	5 persons	8 persons	10 persons

\*; The level of significance is 5%

\*\*; The level of significance is 10%

Judging from Table 2, it was difficult to judge which sample was better in comparison with Example 1 because the contents filled in each sample have themselves perfume. However, the difference in quality of packaging material between two samples becomes distinctive as time had passed, and it was judged that the in-

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vented sample 1 was remarkably superior to the compared sample 1. In the upper block of Table 2, the sentence "5 persons were correct" means that five persons among ten persons could distinguish the invented sample 1 from the compared sample 1 after they smelled each sample. In the lower block of Table 2, the wording "5 persons" means that five persons among ten persons felt that the level of smell on the invented sample 1 was better than that of the compared sample 1.

Example 3

Preservation test was performed with reference to each sample mentioned below into which vodka or rum was filled. If vodka or rum at a high density of alcohol is filled into a bag-in-box to be preserved, evil smell is more or less attached to the contents of the solution in a bag made of synthetic resin. Therefore, it is necessary to select a bag which evil smell is little attached to the contents or no evil smell is attached thereto in order not to deteriorate the quality of the contents to maintain the taste of the contents. The vodka and rum used for the experiments are shown in Table 3.

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Table 3

Brand	Density of alcohol (°GL)	Volume (ℓ)	Container
Kirin-seagram vodka	50	18	20ℓ back-in-box
Myersconfectionery rum	40 & 55	20	20ℓ back-in-box

Bags for accommodating the vodka and rum are shown in Table 4.

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Table 4

Maker	Sample	Volume (ℓ)	Structure of film and thickness
Kirin Beer	Invented sample 1	20	PET (100μ)
F company	Compared sample A	20	Single layer PE (0.12 - 0.3 mm)
T company	Compared sample B	20	PE (75 μ) + PE (35 μ)/K-NY(15μ)/PE(55μ)→Side of contacting contents
D company	Compared sample C	20	OPP(20μ)/KON(15μ)/PE(40μ)+PE(40μ)→Side of contacting contents
	Compared sample D	20	OPP(20μ)/KON(15μ)/PE(40μ)+EVA/PE(90μ)→Side of contacting contents

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Nicolay Vodka of 50°GL and Myersconfectionary Rum of 55°GL were filled into each sample. The quantity of contents was 300 mL. The each sample was then left in atmosphere for 1, 2 and 3 months and, thereafter, each content was diluted to 20°GL with pure water. Further, appearance and smell of each sample was observed and detected. These results are shown in Table 5 and Table 6.

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Table 5

Sample	Appearance
Invented sample 1	Colorless, transparent
Compared sample A	Colorless, transparent
Compared sample B	Colorless, transparent
Compared sample C	Colorless, transparent
Compared sample D	Cloudy

Table 6

Sample	Level of evil smell						Kind of evil smell	
	Nicolay vodka			Mayersconfectionery rum				
	One month after	Two months after	Three months after	One month after	Two months after	Three months after		
Invented sample 1	-	-	-	-	-	-	None	
Compared sample A	+++	++++	+++++	++	+++	++++	Smell of plastic	
Compared sample B	+	++	+++	±	+	++	Sweet smell of plastic	
Compared sample C	+	+	++	±	± ~ +	+	Smell of plastic	
Compared sample D	++++	+++++	++++	++	+++	++++	Sweet smell of plastic	

\* Level of evil smell

-...no feel, ±...slightly feel, +...clearly feel, ++...moderate-strongly feel,

+++...strongly feel, +++...very strongly feel, +++++...extreme-strongly feel

According to the result of Tables 5 and 6, the compared sample D is not suitable for a container for the above drinks because the compared sample D becomes whitely cloudy when it is diluted to 20 °GL with pure water. Further, in four kinds of bag-in-boxes except the compared sample D, the table 6 shows the order in which the level of sticking of evil smell becomes below.  
 Invented sample 1 <Compared sample C <Compared sample B & Compared sample A  
 In conclusion, the invented sample 1 made of PET is suitable for a bag-in-box for vodka and rum at a high density of alcohol.

#### Example 4

Further, preservation test for beer was performed in the following manner.

The invented sample 1 (20l) made of PET and a compared sample S (20l) made of stainless steel (tank) were prepared to be filled with draft beer including CO<sub>2</sub> of 0.5%. These samples were preserved at a temperature of 20°C for various periods to observe the contents therein periodically. This time, the triangle test was conducted by thirteen persons to distinguish invented sample 1 and compared sample S. The result is shown in Table 7.

Table 7

	One week after	Two weeks after	Four weeks after	Eight weeks after
5 Distinction between invented sample 1 and compared sample S	4 persons among 13 were correct	5 persons among 13 were correct	4 persons among 13 were correct	8 persons among 13 were correct
10 Person who felt that compared sample S was better than invented sample 1	2 persons	3 persons	2 persons	8 persons

\*; The level of significance is 5%.

20 Judging from the Table 7, there was no difference in quality of beer during four weeks between the two samples 1 and S. Beer in the invented sample 1 was slightly deteriorated eight weeks after. However, draft beer in a large container (above 7ℓ) is normally consumed within four hours after filling. Therefore, the invented sample 1 can be practically used for draft beer.

25 A bag for a bag-in-box according to this invention has a strength larger than a conventional one. Its gas barrier property is also much better than that of a conventional one. An ability of deodorization and keeping perfume was much improved. Smell component hardly attaches to the bag and smell component of the contents in the bag hardly disperses outward through the wall of the bag. A conventional bag-in-box was a laminated container mainly made of polyethylene with a plurality of layers. However, since the bag according to this invention is made of a single layer of polyethylene telephthalate resin, it is suitable for recycle treatment in comparison with the conventional bag-in-box. In these advantageous effects of the bag according to this invention, since smell component of the bag does not attach to the contents therein, the bag according to this invention can be adapted for food and drinks which are apt to be deteriorated by attachment of smell component except contents. For example, this bag-in-box can be used for mineral water, fruit syrup, flavor undiluted solution, rum and vodka for material of cakes and soups. Further, since this bag-in-box has a good pressure resistance and gas barrier property, it can be adapted for drinks including carbon dioxide such as beverage and beer. Further, since this bag-in-box is rather inexpensive, it is suitable for a disposable bag for drinks.

#### Claims

- 40 1. A bag accommodated in an outer box of a bag-in-box, which is made of film-like material so as to be collapsible to push out contents therein such as beverage syrups and liquid foods, comprising a collapsible bag body made of polyester resin at a thickness of 20 - 150 µm, preferably 50 - 126µm, and a mouthpiece provided on the bag body for passing the contents therethrough.
- 45 2. A bag according to claim 1, wherein said mouthpiece is made of polyester resin.
3. A bag according to claim 1, wherein said mouthpiece is formed as an assembled body in which metal is combined with polyester resin.
- 50 4. A bag according to claim 1, wherein said bag body is made of a single layer of polyester resin.
5. A bag according to claim 1, wherein said bag body is made of an intermediate layer of polyamide resin between laminated layers of polyester resin.
- 55 6. A bag according to claim 1, wherein the polyester resin includes thermal stabilizer, weather resistant stabilizer, antistatic agent, lubricant, mould releasing agent, inorganic filler, pigment, pigment dispersing agent or die.

7. A bag according to claim 1, wherein said polyester resin comprises polyethylene terephthalate resin, polyethylene naphthalate resin, and/or copolymerized polyester resin.
- 5 8. A bag according to claim 7, wherein said polyethylene telephthalate resin includes: terephthalic acid or ester-forming derivative of the terephthalic acid; and ethylene glycol or ester forming derivative of the ethylene glycol.
- 10 9. A bag according to claim 7, wherein said polyethylene naphthalate resin includes: ethylene-2,6 naphthalene component unit derived from 2,6-naphthalene dicarboxylic acid and ethylene glycol.
- 10 10. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit and isophthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit.
- 15 11. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit and 2,6-naphtalene dicarboxylic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit.
- 20 12. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit and adipic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit.
- 25 13. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and diethylene glycol component unit.
14. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit and isophthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and neopentyl glycol component unit.
- 30 15. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit including terephthalic acid component unit; and dihydroxy compound structural unit including ethylene glycol component unit and cyclohexone-dimethanol unit.
- 35 16. A bag according to claim 7, wherein said copolymerized polyester resin comprises: dicarboxylic acid structural unit, dihydroxy compound structural unit and multifunctional hydroxy compound structural unit including at least three hydroxy groups.

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FIG. 1

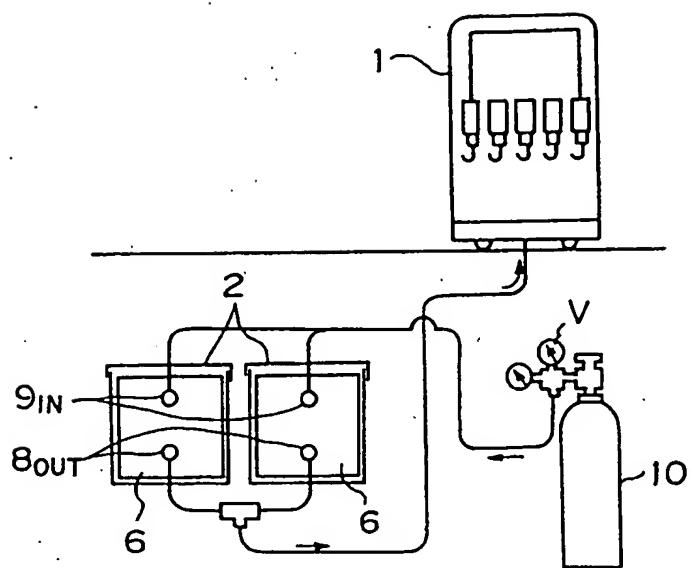


FIG. 2

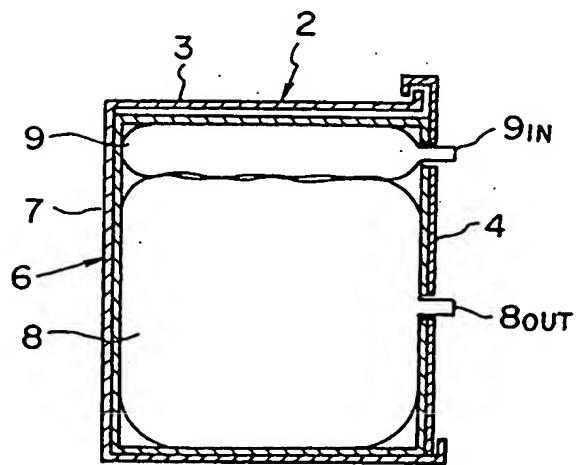


FIG.3

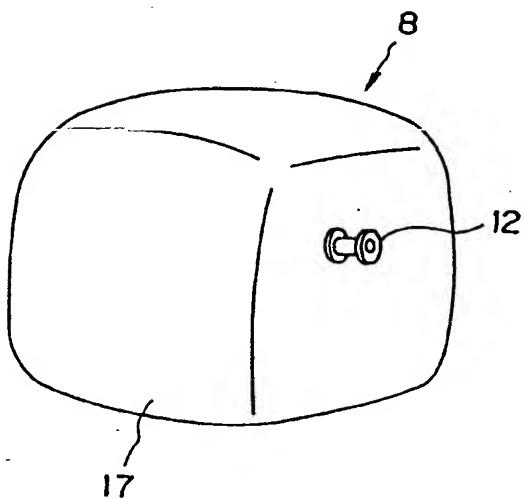
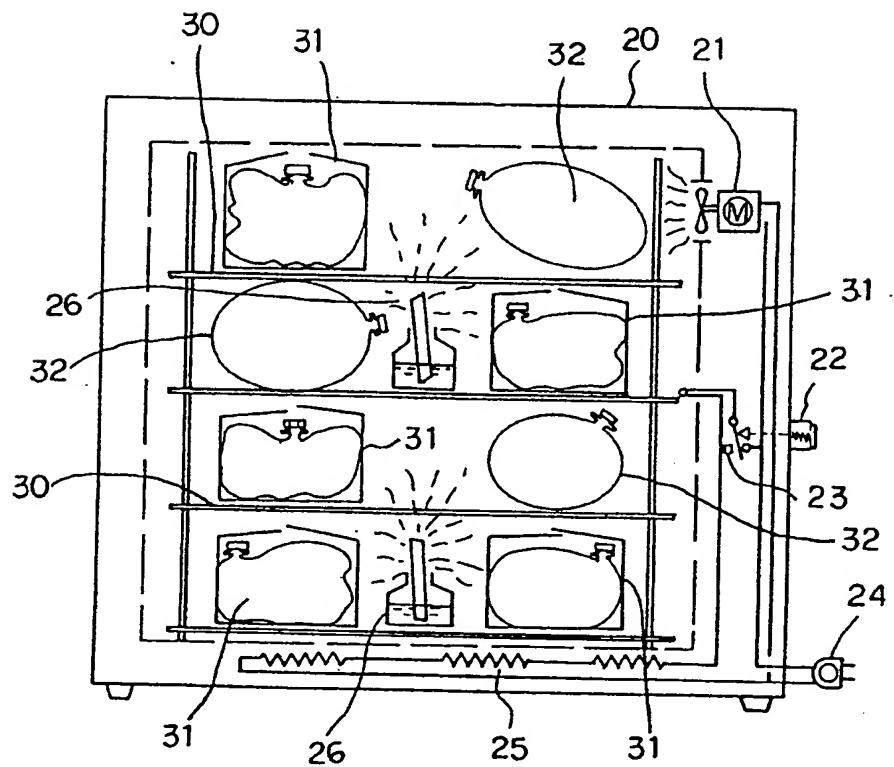


FIG.4





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## EUROPEAN SEARCH REPORT

Application Number

EP 93306391.9

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
X	<u>US - A - 4 704 325</u> (CROCKER) * Column 6, lines 5-7; column 3, lines 42-58; column 5, lines 19-23, 35-45; abstract --	1,5-15	B 65 D 77/06
X	<u>US - A - 4 971 863</u> (HART) * Column 7, lines 42-44; column 6, lines 56-60; column 4, lines 30-45; column 7, lines 4-13 *	1,5-15	
X	<u>US - A - 5 035 945</u> (HART) * Column 7, lines 5-7; column 6, lines 20-24; column 3, lines 50-67 *	1,5-15	
A	DATABASE WPIL, no. 82-45087E DERWENT PUBLICATIONS LTD., London, GB & JP-A-57-068 360 --	1,4	TECHNICAL FIELDS SEARCHED (Int. Cl.5)
A	<u>US - A - 4 696 840</u> (McCULLOUGH et al.) * Claims 1,2,9 *	1,4,7, 8,10	B 32 B 27/00 B 65 D 77/00
A	<u>US - A - 4 851 272</u> (KNOX, III et al.) * Claims 19,33; abstract *	1,7,8	
The present search report has been drawn up for all claims			
Place of search <b>VIENNA</b>	Date of completion of the search <b>18-11-1993</b>	Examiner <b>PUSTERER</b>	
<b>CATEGORY OF CITED DOCUMENTS</b>		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application I : document cited for other reasons & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

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